Answer on Question 85719 in General Chemistry
.a)When aqueous solutions of $\mathrm{Pb}(\mathrm{NO} 3) 2$ and CuCl 2 are mixed we can see the precipitate which is formed according the equation:
.b) $\mathrm{Pb}\left(\mathrm{NO}_{3}\right)_{2}(\mathrm{aq})+\mathrm{CuCl}_{2}(\mathrm{aq})=\mathrm{PbCl}_{2}(\mathrm{~s})+\mathrm{Cu}\left(\mathrm{NO}_{3}\right)_{2}(\mathrm{aq})$
$\mathrm{Pb}^{2+}+2 \mathrm{Cl}^{-}=\mathrm{PbCl}_{2}$

Answer provided by www.AssignmentExpert.com

