Answer on Question 85719 in General Chemistry

.a)When aqueous solutions of Pb(NO3)2 and CuCl2 are mixed we can see the precipitate which is formed according the equation:

.b) 
$$Pb(NO_3)_2$$
 (aq)+ $CuCl_2$  (aq)=  $PbCl_2$ (s) + $Cu(NO_3)_2$ (aq)

$$Pb^{2+} + 2 Cl^{-} = PbCl_{2}$$

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