Answer on Question 84698 in General Chemistry.
$\mathrm{M}\left(\mathrm{Cl}_{2}\right)=70.9 \mathrm{~g} / \mathrm{mol}$
$. \mathrm{m}\left(\mathrm{Cl}_{2}\right)=150 \mathrm{~g}$
. $\mathrm{n}\left(\mathrm{Cl}_{2}\right)=$ ?
$. \mathrm{n}=\frac{m}{M}=\frac{150}{70.9}=2.1 \mathrm{~mol}$

Answer provided by www.AssignmentExpert.com

