

if the molecular surface of two myoglobin molecule are uprated by a distance of 20 Angstroms diameter of each myoglobin is 40 angstroms diploe moment of myoglobin is 170 D at 20 angstrom and are collinear present then calculate the energy of interaction between two molecules?

Answer: since there is collinearity, and the compound itself is a high-molecular compound in terms of high molecular weight, these two factors favor the calculation of the interaction energy as follows: $W_{\text{ind}} = \frac{-\alpha p^2}{2\pi\epsilon\epsilon_0 r^6}$, where α -polarizability of the molecule or atomic group; p - dipole moment; r is the distance between molecules, in our case it is 20 Angstroms; ϵ - relative dielectric constant of the medium; $\epsilon_0 = 8.85 \times 10^{-12}$ F/m – electric constant.

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