V_1 = 450 mL

P_1 = 756 torr

T_1 = -10 °C = 14 °F

P_2 = 752 mmHg = 752 torr

T_2 = 98.6 °F

$$\frac{P1*V1}{T1} = \frac{P2*V2}{T2}$$

$$V_2 = \frac{P1*V1*T2}{T1*P2} = \frac{756torr*450mL*98.6°F}{14°F*752torr} = 3186.14 \text{ mL}$$