## Question \# 83230, answer

How many grams of NaCl are required to make 500 g of a $0.9 \%$ (w:w) solution?

## Answer:

The label $0.9 \%(\mathrm{w} / \mathrm{w})$ indicates that 0.9 g of NaCl is present for every 100 grams of solution (by definition). Therefore, amount of NaCl required to make 500 g is $0.9 \mathrm{~g} \times(500 \mathrm{~g} / 100 \mathrm{~g})=$ 4.5 g

## Answer provided by www.AssignmentExpert.com

