## Answer on Question 82528 in General Chemistry

The highest boiling point in 0.05 M KBr because KBr is a strong electrolyte which dissociates by the equation

K Br = 
$$K^+$$
 +B $r^-$   
. $\Delta$ t= i ×  $C_m$  ×E i=2  
 $E_{H_{2O}}$ =0.52×2×0.05=0.052

 $.t_b$ =  $t_{H_{2\ O}}$ + $\Delta$ t=100°+0.052°=100.052°C

Answer provided by www.AssignmentExpert.com