A block of gold measures $2.5 \mathrm{~cm} \times 1.5 \mathrm{~cm} \times 6.5 \mathrm{~cm}$ the density of gold is $19.3 \mathrm{~g} / \mathrm{cm}$ cubed. calculate the mass of this block of gold

Solution:
1.V=S×H;
$\mathrm{V}=2.5 \times 1.5 \times 6.5=24.375 \mathrm{~cm} 3 ;$
2. 19.3 grams- 1 cm 3
x grams- 24.375 cm3
$x=19.3 \times 24.375=470.44$
m (gold) $=470.44$ grams
Answer:m(gold)=470.44 grams.

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