

Question #82504, Chemistry / General Chemistry | for completion

Which of the following explains why nonmetals are typically harder and more brittle than metals?

Answer:

There are not explains, but I can Answer this question.

Nonmetals are typically harder and more brittle than metals. Because such bonds could be formed between metal atoms that have low electronegativities and do not attract their valence electrons strongly. Nonmetals are not.

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