

Question #81910, Chemistry / General Chemistry

Question:

How many oxygen atoms are there in 7 formula units of $\text{Ca}_3(\text{PO}_4)_2$?

Solution:

One molecule of $\text{Ca}_3(\text{PO}_4)_2$ contains $4 \times 2 = 8$ oxygen atoms.

As a result, 7 formula units of $\text{Ca}_3(\text{PO}_4)_2$ contain $8 \times 7 = 56$ oxygen atoms.

Answer:

56 oxygen atoms

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