Enter the balanced ionic and net ionic equation of: H2SO3(aq)+2NH3(aq) --> (NH4)2SO3 (aq)

Solution:

H2803(ap) + 2 M3(ap) -> (M4), PO3(ap) H2803 (aq) + 2 M3 (aq) -> 2NH3 + + 803 (aq) Strong electrolytes, dissociated virtually completely, are recorded in the form of ions Low-dissociated (weak electrolytes), sport soluble and gaseous substances are recorded in the form of molecules.

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