

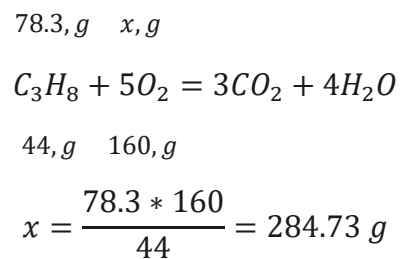
### Question #81323

How many grams of O<sub>2</sub>(g) are needed to completely burn 78.3 g of C<sub>3</sub>H<sub>8</sub>(g)?

Answer:

The mass of oxygen, which is needed to burn 78.3 g of propane, is equal to 284.73 g.

According to the chemical equation, the mass of oxygen, which is needed to burn 78.3 g of C<sub>3</sub>H<sub>8</sub>, is equal to:



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