## Answer on Question #79705, Chemistry / General Chemistry

How many neutrons are in this isotope  ${}^{14}_{6}{\cal C}$ ?

## Solution

14 – mass number

6 – number of protons

$$N_n = 14 - 6 = 8$$

## **Answer**

**8** neutrons are in the isotope  ${}^{14}_{6}\mathcal{C}$ .

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