Answer on Question 78229 in General Chemistry

The highest boiling points will have 0.75 M solution of $CuCl_2$ because it's strong electrolyte which is completely dissociated in aqueous solution according the equation

 $CuCl_2 = Cu^{2+} + 2Cl^-$

The general concentration of ions is $3 \times 0.75 = 2.25$ M