

Answer on Question #78038 – Chemistry – General Chemistry

\_\_\_\_\_ will have a higher entropy than 50.0 g CO<sub>2</sub> (g) at 56°C.

- A. 25.0 g CO<sub>2</sub> (g) at 56°C
- B. 50.0 g CO<sub>2</sub> (g) at 50°C
- C. 50.0 g CO<sub>2</sub> (l) at 56°C
- D. 55.0 g CO<sub>2</sub> (l) at 56°C
- E. 55.0 g CO<sub>2</sub> (g) at 56°C

**Solution:**

- E. 55.0 g CO<sub>2</sub> (g) at 56°C

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