

_____ has a ΔH_f of zero.

- A. Br⁻
- B. Na⁺
- C. H₂O
- D. CO₂
- E. None of the Above

Solution:

The standard formation enthalpy of simple substances is assumed to be zero, moreover the zero value of the formation enthalpy referring to the aggregate state stable at T = 298.15 K.

Answer: E. None of the Above.

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