

## Answer on Question #78032 – Chemistry – General Chemistry

Which of the following statements is correct?

- A. In endothermic reactions, heat is released.
- B. In exothermic reactions, heat is absorbed.
- C. If  $\Delta G < 0$ , then the reaction is consistent with the Second Law of Thermodynamics.
- D. If  $\Delta G > 0$ , then the reaction is consistent with the Second Law of Thermodynamics.
- E. All of the Above

**Solution:**

- C. If  $\Delta G < 0$ , then the reaction is consistent with the Second Law of Thermodynamics.

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