

Answer on Question #78026, Chemistry / General Chemistry

Question:

The symbol _____ stands for free energy, according to the relationship free energy = change in enthalpy – (temperature × change in entropy).

- A. ΔE
- B. ΔG
- C. ΔH
- D. ΔS
- E. ΔT

Answer:

The law described in the task: $\Delta G = \Delta H - T \cdot \Delta S$, so the answer:

- B. ΔG**