

Answer on Question #77316 - Chemistry - General Chemistry

Question:

A proton is Accelerated to one - tenth of the velocity of light. suppose its velocity can be measured with a precision of +- 1% , what must be the uncertainty in its position (in nm) (mass of proton= 1.673×10^{-27})

Solution:

$$\text{Velocity} = 3 \times 10^8 / 10 = 3 \times 10^7 \text{ m/sec};$$

$$\Delta V = V \times 1/100 = 3 \times 10^5;$$

$$\Delta x = h / 4\pi m \Delta V = 6.626 \times 10^{-34} / (4\pi \times 1.673 \times 10^{-27} \times 3 \times 10^5) = 1.05 \times 10^{-13} \text{ m}.$$

Answer: $1.05 \times 10^{-13} \text{ m}$.