

Answer on Question #77125 – Chemistry – General Chemistry

15.25 mL of 2.10 M CaCl_2

Solution:

$$n(\text{CaCl}_2) = C \times V = (2.10 \text{ mol/L}) \times (0.01525 \text{ L}) = 0.0320 \text{ moles}$$

Answer provided by AssignmentExpert.com