

## Answer on Question #76933 - Chemistry - Physical Chemistry

Question:

If a certain metal was irradiated by using two different light radiations of frequency  $x$  and  $2x$  the kinetic energies of the ejected electrons are  $y$  and  $3y$  respectively the threshold frequency of the metal will be

**Solution:**

$$E_k = h\nu - h\nu_0;$$

$$h\nu - h\nu_0 = y;$$

$$h_{2x} - h\nu_0 = 3y;$$

$h\nu_0$  will be same as it is work function and it is a property of a metal;

$$h(\nu - \nu_0) = y \quad h(2\nu - \nu_0) = 3y ;$$

$$3\nu - 3\nu_0 = 2\nu - \nu_0;$$

$$\nu = 2\nu_0;$$

$$\nu_0 = \nu/2.$$

**Answer:**  $\nu/2$

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