Answer on Question #76215 - Chemistry - Physical Chemistry

Question:

Briefly explain the laws of osmotic pressure

Solution:

Want Hoff Law:

The osmotic pressure of the solution is equal to the pressure that would create the dissolved substance if it was in a gaseous state and at the same temperature occupied a volume equal to the volume of the solution.

$P = C*R*T$

$P*V = n*R*T$