## Question 1

name three commercial uses for alum?

## Answer:

- 1. Construction material: railing, rods, doors, building wire, sheathing, roofing.
- 2. Transportation (automobiles, aircraft, trucks, railway cars, marine vessels, bicycles, spacecraft, etc.) as sheet, tube, and castings.

Properties of aluminium that allow these uses (1 and 2): resistance to corrosion, high strength to weight ratio, it's easy to paint, form and connect to other materials.

- Heat sinks for CPU's and graphic processors.
  Properties of aluminium that allow this use: heavy thermal, corrosion, biofuoling resistance (for aluminium alloys) and thermal conductivity.
- Long-distance power lines.
  Properties of aluminium that allow this use: high conductivity (being mixed with copper) and light weight.
- Rolled sheet Products.
  Properties of aluminium that allow this use: the ability to be rolled into super thin foil for food packaging or thicker sheets for cans.

## Question 2

which happens when 3M KOH solution is added to aluminum foil what gas is released?

## Answer

Aluminuim foil is made out of aluminiun which reacts with alkali (KOH) producing complex salt and hydrogen. Aluminium foil is dissolved and hydrogen is released.

 $2AI(s) + 2KOH(aq) + 6H_2O(I) = 2KAI(OH)_4(aq) + 3H_2(g)$