

Answer on Question #72028, Chemistry / General Chemistry

Question:

If 12.0 g of KCl is dissolved in 88.0 g of water, what is the mass/mass percent KCl?

Solution:

Mass percent = (Mass of solute / Mass of solution) · 100%

So:

Mass percent KCl = (12.0 / 100.0) · 100% = 12%

Answer:

12%