

Answer to the Question #67590, Chemistry / General Chemistry

Calculate the molarity of each solution. 38.2 g of KCl in 0.895 L of solution

$$C = \frac{m}{M \cdot V}$$

$$M(\text{KCl}) = 74.5$$

$$C = \frac{38.2}{74.5 \cdot 0.895} = 0.573 \text{ mol/L}$$

Answer provided by [www.AssignmentExpert.com](http://www.AssignmentExpert.com)