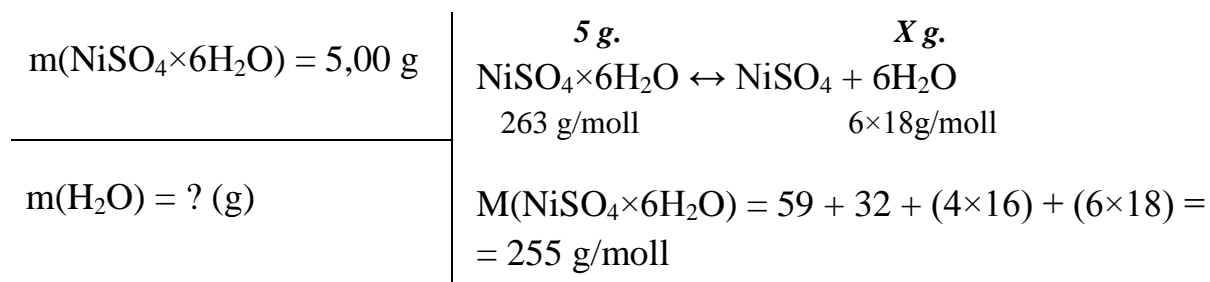


Question #65065, Chemistry / General Chemistry

Task:



$$263 \text{ g. (NiSO}_4 \times 6\text{H}_2\text{O}) - 108 \text{ g. (H}_2\text{O)}$$

$$5 \text{ g. (NiSO}_4 \times 6\text{H}_2\text{O}) - X \text{ g. (H}_2\text{O)}$$

$$X = \frac{5 \times 108}{263} = 2.053 \text{ g.}$$

Answer: 2.053 grams of water are present.

Answer provided by <https://www.AssignmentExpert.com>