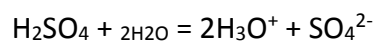


Question #64982, Chemistry / General Chemistry

Calculate the $[H_3O^+]$ for 0.0058 M H_2SO_4 . Answer in unit M

Answer:



$$[H_3O^+] = 2 \times [H_2SO_4] = 2 \times 0.0058 M = \mathbf{0.0116 M}$$

Answer provided by <https://www.AssignmentExpert.com>