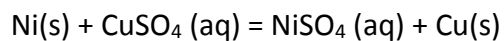**Answer on the Question #63758, Chemistry / General chemistry**

In the correct equation for the reaction of a strip of Ni(s) dipped in 1M CuSO4, the sum of the simplest whole-number coefficients for all products is:

Answer:

$$Ni(s) + CuSO_4 (aq) = NiSO_4 (aq) + Cu(s)$$

The sum of the whole-number coefficients for all products equal to:

$$N = 1 \ NiSO_4 (aq) + 1 \ Cu(s) = 2$$