## Question #60421, Chemistry, Other

ANSWER THE QUESTIONS:

- 1. Are particles in all forms of matter the same? Yes No
- 2. How are the particles in liquids different from the particles in solids?
- 3. Why can gas fill any space?
- 4. Compare the properties of matter based on (shape and volume, space between the particles, force of attraction).

## Answer:

- 1. No
- 2. The particles in solids are closer to each other and bounded by stronger bounds
- 3. The gas can fill any space because the bounds between its particles are weak.
- 4. There are three forms of matter:
  - Gas (week bounds between the particles, no shape, can fill any volume);
  - Liquid (comparatively stronger bounds between the particles, holds its volume, no shape);
  - Solid (stronger bounds between the particles, holds its shape and volume).

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