## Answer on Question # 60315 - Chemistry - General Chemistry

On warming crystals of NaCl with conc H<sub>2</sub>SO<sub>4</sub>, A SALT IS FORMED and a gas is produced

1.name the gas

- 2. state 2 physical properties of the gas produced
- 3. write a balanced equation for the reaction

## Solution

Reaction between sodium chloride and concentrated sulfuric acid leads to a salt sodium bisulfate and hydrogen chloride gas:

 $NaCl(s) + H_2SO_4(aq) = HCl(g) + NaHSO_4(aq).$ 

The name of the gas is hydrogen chloride;

The physical properties of the gas produced: colorless, highly soluble in water.