

Question #59501, Chemistry / General Chemistry

A patient's blood test shows that she has elevated levels of LDL (Low-density lipoproteins) at 4.14mM. How many moles of LDL are there in 8.00 ml of blood

Answer:

The amount of substance is evaluated in the following manner:

$$v = V \times C = 8 \times 10^{-3} L \times 4.14 \times 10^{-3} \frac{mol}{L} = 3.312 \times 10^{-5} mol$$